



Detailed Work Instruction for Electroplating Non-Conductive Parts

This instruction describes the professional electroplating of non-conductive substrates such as glass, plastics, GRP/CFRP, and 3D-printed parts. It fully integrates your provided process description and expands it with technical details, best practices, and safety guidance.

1. Scope and Objective

The objective is to make non-conductive surfaces electroplatable by applying a conductive coating (conductive paint) and subsequently building up a mechanically stable, glossy copper layer, followed optionally by further metallic layers (nickel, silver, gold, etc.).

2. Materials and Equipment

2.1 Substrates

- Glass
- Plastics (ABS, PLA, PETG, PA, PC, etc.)
- GRP / CFRP
- 3D-printed parts (FDM, SLA, SLS)

2.2 Chemicals and Consumables

- Copper conductive paint or silver conductive paint
- Acid copper electrolyte (bath copper or PP copper)
- Optional: 10% sulfuric acid (H_2SO_4)
- Degreasing agent (isopropyl alcohol, acetone* depending on plastic)
- Sandpaper grit 400–600 (surface prep), 2000–3000 (activation)
- Deionized water

2.3 Electroplating Equipment

- Regulated DC power supply (stable output)
- Electrolyte tanks (chemically resistant)
- Electrodes:
 - Bath plating: copper plate electrodes

- Brush / tampon plating: rod electrode Ø 6 mm
- Suspension and electrical contacts (copper wire, clamps)
- Gentle agitation or solution movement (recommended)

2.4 Personal Protective Equipment (PPE)

- Chemical-resistant gloves
 - Safety goggles
 - Acid-resistant apron
 - Adequate ventilation / exhaust
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3. Substrate Preparation

3.1 Mechanical Surface Preparation

1. Lightly roughen the surface to improve adhesion:
 - Smooth plastics / glass: sandpaper grit 400–600
2. Remove all sanding dust thoroughly.

3.2 Degreasing

1. Thoroughly degrease the surface using a suitable solvent.
2. Rinse with deionized water.
3. Allow to dry completely.

Note: Residues from release agents (especially on 3D-printed parts) are a very common cause of adhesion failure.

4. Application of Conductive Paint

All non-conductive surfaces can be electroplated as follows:

4.1 Selection of Conductive Paint

Silver conductive paint

- Higher cost
- Very reliable and high conductivity
- No activation required

Copper conductive paint

- More economical
- Forms an oxide/patina after drying
- Requires activation prior to plating

4.2 Application

1. Apply the conductive paint very thinly and evenly:
 - Preferably by airbrush
 - Alternatively by spray can
2. Several thin coats are better than one thick coat.

3. Define and reinforce the electrical contact point.

4.3 Drying

- **Allow the conductive paint to fully cure:**
 - **Not only surface-dry**
 - **Must be well bonded to the substrate**
- **Follow manufacturer curing times (often several hours up to 24 h).**

Important: Insufficiently cured conductive paint may detach in the electrolyte.

5. Activation of Copper Conductive Paint

Because copper conductive paint oxidizes after drying, activation is required:

5.1 Activation in Copper Electrolyte (recommended)

- **Immerse the part in acid copper electrolyte for approx. 20 minutes without current**
- **The acid removes the oxide layer**

5.2 Alternative Activation Methods

- **Brief treatment with 10% sulfuric acid**
- **Very light sanding using 2000–3000 grit sandpaper**

After activation:

- **Rinse with deionized water**
 - **Avoid touching the surface (finger grease!)**
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6. Copper Electroplating (Base Build-Up)

6.1 Plating Methods

- **Bath plating (uniform, thicker deposits)**
- **Brush / tampon plating (localized areas, small parts, repairs)**

6.2 Bath Plating Setup

- 1. Select a tank large enough to fully immerse the part.**
- 2. Use 1–4 copper plate electrodes, depending on tank size.**
- 3. Distance between electrodes and part: approx. 10 cm.**
- 4. Ensure secure and conductive electrical contact to the part.**

6.3 Process Parameters (Guideline Values)

- **Start with low current density, increase slowly**
- **Voltage depends on bath chemistry and geometry**
- **Temperature according to electrolyte supplier**
- **Plating time: long and intensive until a**
 - **closed**
 - **glossy**

- mechanically stable copper layer is achieved

The goal is not only conductivity, but sufficient layer thickness and adhesion.

7. Further Plating / Barrier Layers

After copper build-up, the following options are available:

7.1 Hard Barrier Layers (recommended)

- Nickel
- Palladium
- Cobalt

These layers prevent diffusion, oxidation, and significantly improve wear resistance.

7.2 Direct Finishing on Copper

- Silver plating
- Gold plating
- Black nickel plating

Copper can be plated directly, but a barrier layer greatly improves long-term stability.

8. Suitable Electrodes

- Brush / tampon plating: rod electrode Ø 6 mm
- Bath plating: copper plate electrodes

Electrodes should be cleaned regularly to avoid passivation.

9. Typical Errors and Practical Notes

- Insufficient degreasing → poor adhesion
 - Conductive paint applied too thick → peeling
 - Excessive current density → rough or burnt deposits
 - Poor electrical contact → uneven plating
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10. Safety Notes

- Acids and electrolytes are corrosive
- Never add water to acid
- Rinse splashes immediately with plenty of water
- Store chemicals properly labeled and out of reach of children

Equipment:

Bezeichnung
Kupfer- oder Silberleitlack: https://real-gold.de/Produkte/Leitlacke-nicht-leitende-Flaechen/
Saures Kupfer https://real-gold.de/Kupfer-sauer-Glanzkupfer-Elektrolyt/BMG-095M
PP Kupfer: https://real-gold.de/PP-Kupfer/BMG-130M
25 A Netzteil regelbar https://real-gold.de/Galvanikgeraet-Laborgeraet-regelbar/BMG-25A
Kabel PLUS https://real-gold.de/Anschlusskabel-PLUS-mit-Bananenstecker/BMG-025
Kabel MINUS mit Klemme https://real-gold.de/Anschlusskabel-MINUS-mit-Stecker-und-Klemme/BMG-026
Griff https://real-gold.de/Handgriff-Anodenhalter/BMG-027
Edelstahlelektroden dick 6mm zum Entfetten und Aktivieren https://real-gold.de/Edelstahl-Elektrode-6-mm/BMG-007
Graphitelektrode https://real-gold.de/Graphit-Elektrode-6mm/BMG-010
Kupferelektrode https://real-gold.de/Kupfer-Elektrode-6mm/BMG-011
Plattenelektrode Kupfer: https://real-gold.de/Kupfer-Platten-Elektrode/SW10051M
Plattenelektrode Graphit: https://real-gold.de/Graphit-Plattenelektroden/SW10030.2M
Stoffpad bauschig für jedes Elektrolyt 1x https://real-gold.de/Anoden-Stoffpad-bauschig/BMG-015
Becher mit Deckel https://real-gold.de/Aufbewahrungsbecher-fuer-Elektrolyt-und-Elektrode/BMG-023-A
Entfetter https://real-gold.de/Elektrocleaner-Entfetter/SW10002.5M

Aktivator

<https://real-gold.de/Aktivator/BMG-061.3M>

Nickel Free oder Nickel Strike

<https://real-gold.de/Nickel-Free/BMG-091M>

Gold:

<https://real-gold.de/Gold-Elektrolyt-8-Gramm-L.-Gold/BMG-067M>

Betzmann Galvanik – Oberflächenbeschichtung mit 24 Karat Gold-Silber-Platin-Chrom

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